# Stereochemistry Exam Preparation Pack Answer Key- Advanced

## Section A: Find Chiral Centers and Determine R/S

Find the chiral centers in each of these molecules with "alternative uses" and determine *R/S* for each chiral center.

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## Section B: Convert to Fischer Projection

For each of the three molecules below:

- · Label each chiral center as R/S
- Convert the drawing into a Fischer projection
- Draw the other stereoisomers as Fischer projections
- · Indicate which of these stereoisomers is the enantiomer

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• Indicate which stereoisomer(s) are diastereomers



B-1 2,3-Dibromosuccinic acid

no enantiomer (meso!)

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other stereoisomers (both diastereomers)





Fischer projection

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Enantiomer

CI-

Diastereomer (meso!)

B-3 3-Chlorobutan-2-ol

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http://bit.ly/Stereochem-AD-MOC-5



# Section C: Chiral or Achiral Molecules?

C-1 Chiral or achiral molecules? If meso, indicate

chiral







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achiral meso chiral





achiral meso

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## D-1 Draw the enantiomer (+ more)

Your mission is to:

1) identify all chiral centers in Zocor

- 2) Draw the enantiomer
- 3) How many stereoisomers

are possible for Zocor?



Zocor

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*enantiomer 2<sup>7</sup> stereoisomers possible* 

## E-1 Enantiomers, Diastereomers, Constitutional Isomers, or the Same? http://bit.ly/Stereochem-AD-MOC-10

For each pair: Are these molecules enantiomers, diastereomers, the same, or constitutional isomers? Would an equal mixture of these two compounds rotate plane-polarized light?







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E-2 Enantiomers, Diastereomers, Constitutional Isomers, or the Same? http://bit.ly/Stereochem-AD-MOC-11





E-3 Enantiomers, Diastereomers, Constitutional Isomers, or the Same? http://bit.ly/Stereochem-AD-MOC-12







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Section F: Given the name, draw the structure

- a) Draw (2S,3R)-2,3-Difluorohexane using wedge/dash
- b) Draw the diastereomers

E-5

HC





diestere

diastereomer diastereome

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diastereomer (enantiomer)

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### Section G, H, I: Cycloalkanes

**G-1** a) Draw the two *achiral* forms of 1,3,5-Trimethylcyclohexane b) Which is more stable?









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achiral #1

most stable achiral #2 chair form (most stable overall)

most stable chair form

## H-1

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a) Draw the most stable achiral isomer of a cyclohexane with a single fluoro and a single bromo substituent on the ringb) Draw the most stable chiral isomer of a cyclohexane with a single fluoro and a single bromo substituent on the ring





- I-1 a) Draw one version of 1,3-Dimethylcyclohexane that is chiral, and one that is achiral http://bit.ly/Stereochem-AD-MOC-18
  - b) One of these isomers has two conformers of very different energy.
    Draw those two chair conformations.





achiral has conformers of very different energy

## J-1 Draw The Enantiomer (+ more)

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Escitalopram (Celexa):







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• Pure *S* enantiomer shows a specific rotation of  $+120^{\circ}$ . Sven, a worker in the quality control unit, observed a specific rotation of  $-30^{\circ}$  for a test sample. What is the percentage of (*R*) and the percentage of (*S*) in that sample?

62.5 % (R) 37.5 % (S)

## **K-1 Optical Activity**

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An 80:20 mixture of the (R,R) and (S,S) enantiomers of 2,3-dibromobutane has an optical rotation of  $-30^{\circ}$ .

Using these templates, show the stereochemical representation of these compounds, their stereoisomers, and their optical rotations:





## (a chiral amine)

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(+)-mandelic acid (-)-mandelic acid

How are these products related to each other?
 diastereomers

Racemic mixture of mandelic acid: reaction with (+)-Brucine

Draw the two products of the following reaction, clearly showing stereochemistry (it's OK to use " $R_3N$ " for (+)-brucine). Note that (+/–) implies a 1:1 mixture of en-

• How might you exploit this to resolve mandelic acid into its enantiomers? Describe this process briefly

## 1) add (+)-brucine to racemic mixture. 2) recrystallize, and separate crystals (one diastereomer) from mother liquor, which contains other diastereomer. 3) add acid to crystals, re-forming mandelic acid, then extract in separatory funnel using water and organic solvent to separate optically active mandelic acid from the salts (can do the same for the mother liquor)

## **M-1 Chiral Nitrogens**

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Although the nitrogen in the molecule **A** below has four different substituents, the nitrogen does not give rise to a pair of enantiomers. Why not?

Would you expect the nitrogen in molecule **B** to be a chiral center? Why or why not?

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## be separated

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antiomers.





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