

# Stereochemistry Exam Preparation Pack

## Answer Key- Advanced

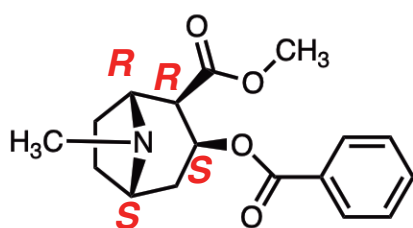
note - all problems can also be found [here](#) (link)

### Section A: Find Chiral Centers and Determine R/S

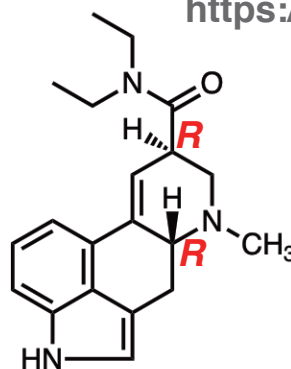
Find the chiral centers in each of these molecules with “alternative uses” and determine *R/S* for each chiral center.

[Link to answer](#)

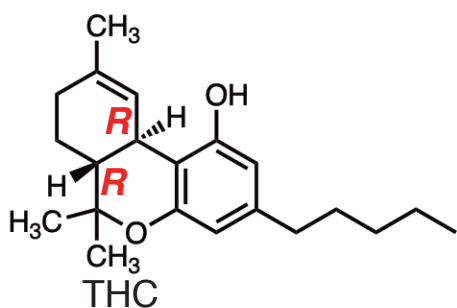
<https://bit.ly/3F2e8Sc>



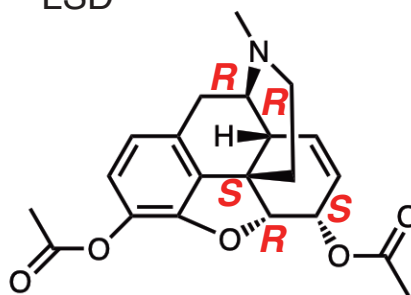
Cocaine



LSD



THC



Heroin

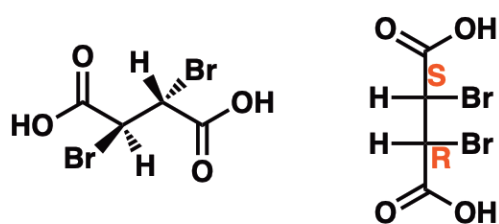
### Section B: Convert to Fischer Projection

For each of the three molecules below:

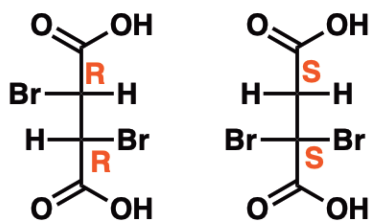
- Label each chiral center as R/S
- Convert the drawing into a Fischer projection
- Draw the other stereoisomers as Fischer projections
- Indicate which of these stereoisomers is the enantiomer
- Indicate which stereoisomer(s) are diastereomers

**B-1** 2,3-Dibromosuccinic acid

<https://bit.ly/2WjbAxm>



*no enantiomer (meso!)*

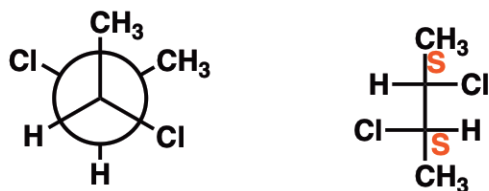


*other stereoisomers (both diastereomers)*

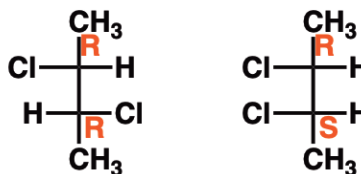


**B-2** 2,3-Dichlorobutane

<https://bit.ly/3AUeGXJ>



*Fischer projection*



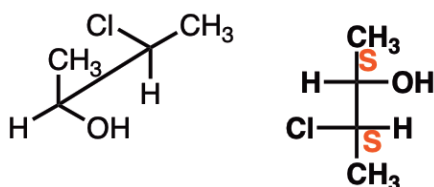
*Enantiomer*

*Diastereomer (meso!)*

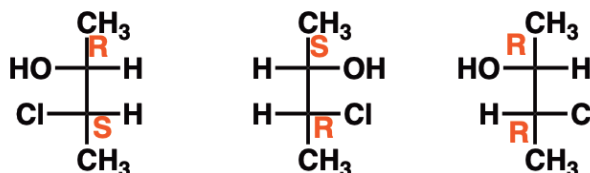


**B-3** 3-Chlorobutan-2-ol

<https://bit.ly/3kKhoJH>



*Fischer projection*



*Diastereomer*

*Diastereomer*

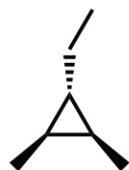
*Enantiomer*



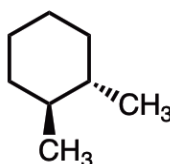
**Section C: Chiral or Achiral Molecules?**

**C-1** Chiral or achiral molecules?  
If meso, indicate

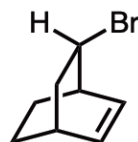
<https://bit.ly/2XWI5lm>



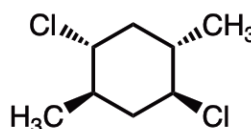
**achiral**  
**meso**



**chiral**



**chiral**

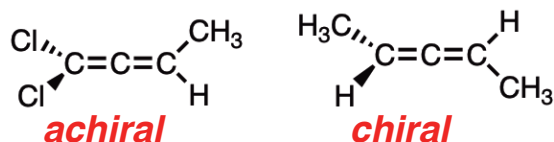


**achiral**  
**meso**



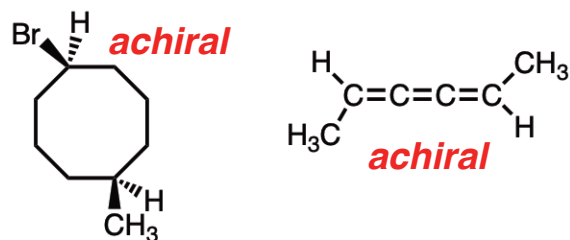
**C-2** Chiral or achiral molecules? If meso, indicate

<https://bit.ly/3CSBO9E>



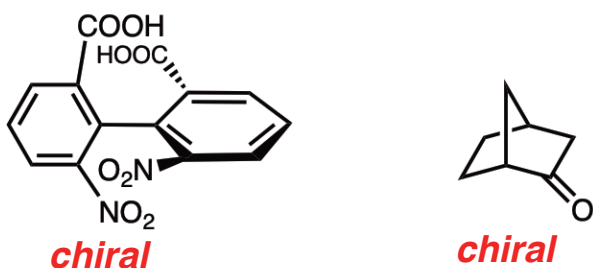
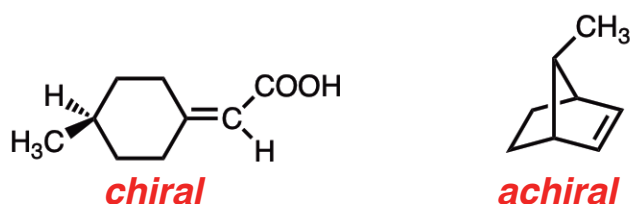
**C-3** Chiral or achiral molecules? Indicate if meso

<https://bit.ly/3zKH64Z>



**C-4** Chiral or achiral molecules? Indicate meso (if present)

<https://bit.ly/3um8DJd>

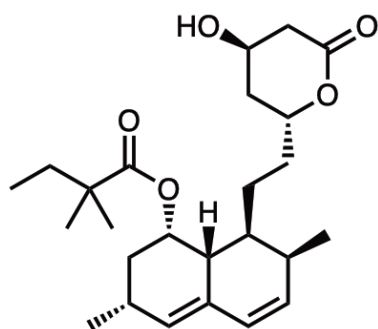


**D-1 Draw the enantiomer (+ more)**

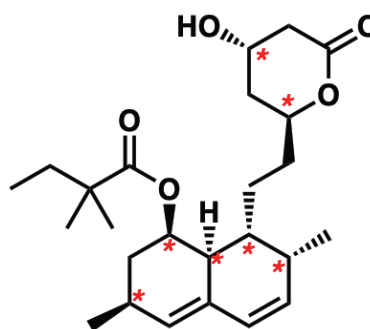
<https://bit.ly/3uhaBdK>

Your mission is to:

- 1) identify all chiral centers in Zocor
- 2) Draw the enantiomer
- 3) How many stereoisomers are possible for Zocor?



**Zocor**



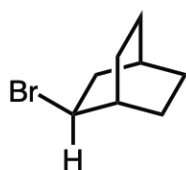
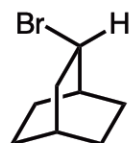
**enantiomer**

**2<sup>7</sup> stereoisomers possible**

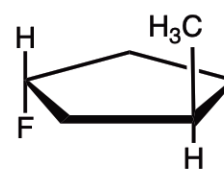
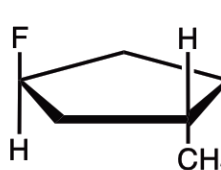
**E-1 Enantiomers, Diastereomers, Constitutional Isomers, or the Same?**

<https://bit.ly/3AX3iuo>

For each pair: Are these molecules enantiomers, diastereomers, the same, or constitutional isomers?  
Would an equal mixture of these two compounds rotate plane-polarized light?



**a) same**  
**b) yes**



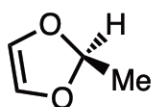
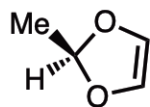
**a) enantiomers**  
**b) no**



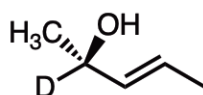
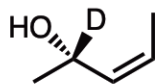
**a) same**  
**b) no**

**E-2** Enantiomers, Diastereomers, Constitutional Isomers, or the Same?

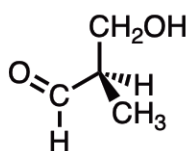
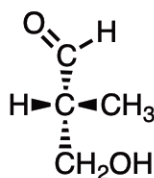
<https://bit.ly/3AQwAe1>



*a) same*  
*b) no*



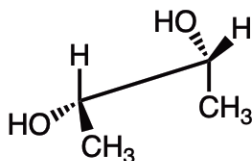
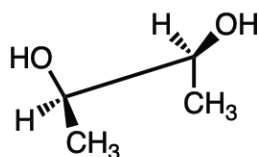
*a) diastereomers*  
*b) yes*



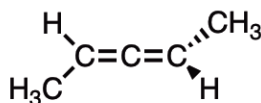
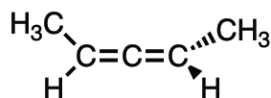
*a) same*  
*b) yes*

**E-3** Enantiomers, Diastereomers, Constitutional Isomers, or the Same?

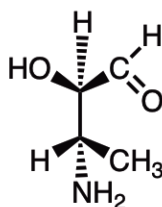
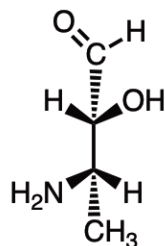
<https://bit.ly/2WIZPX9>



*a) same*  
*b) no*



*a) enantiomers*  
*b) no*

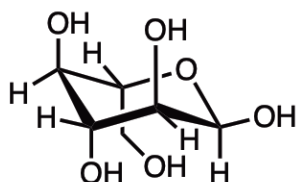
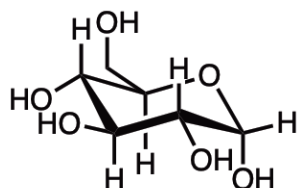


*a) same*  
*b) yes*

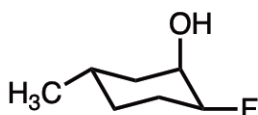
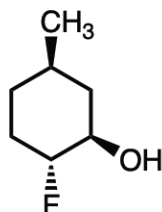


**E-4** Enantiomers, Diastereomers, Constitutional Isomers, or the Same?

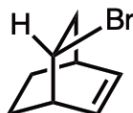
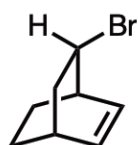
<https://bit.ly/3zVJGpa>



*a) enantiomers*  
*b) no*



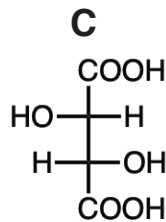
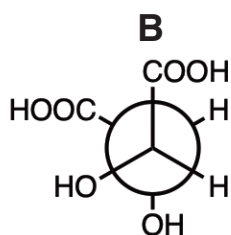
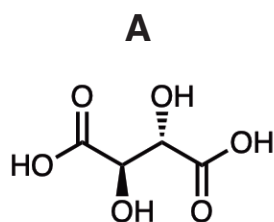
*a) diastereomers*  
*b) yes*



*a) enantiomers*  
*b) no*

**E-5** How are these three molecules (A, B, and C) related to each other?

<https://bit.ly/3zVJOVG>



*A and B: same*  
*A and C: diastereomers*  
*B and C: diastereomers*

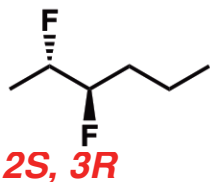


### Section F: Given the name, draw the structure

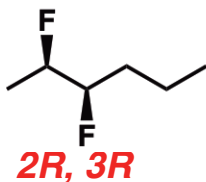
a) Draw (2*S*,3*R*)-2,3-Difluorohexane using wedge/dash

b) Draw the diastereomers

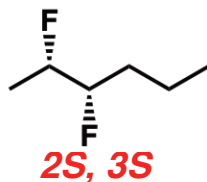
<https://bit.ly/39OwBTT>



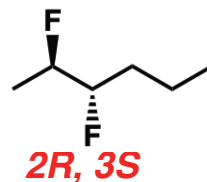
*2S, 3R*



*2R, 3R*



*2S, 3S*



*2R, 3S*

*diastereomer*

*diastereomer*

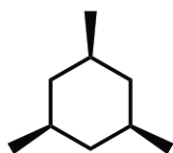
*(enantiomer)*



Section G, H, I: Cycloalkanes

<https://bit.ly/3kPVbtM>

- G-1 a) Draw the two *achiral* forms of 1,3,5-Trimethylcyclohexane  
b) Which is more stable?

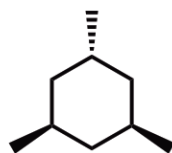


*achiral #1*

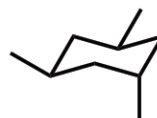


*most stable  
chair form*

*(most stable overall)*



*achiral #2*



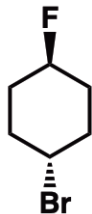
*most stable  
chair form*



H-1

<https://bit.ly/3kPPDzv>

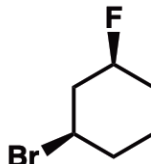
- a) Draw the most stable **achiral** isomer of a cyclohexane with a single fluoro and a single bromo substituent on the ring  
b) Draw the most stable **chiral** isomer of a cyclohexane with a single fluoro and a single bromo substituent on the ring



*achiral*



*most stable  
chair form*



*chiral*

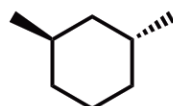


*most stable  
chair form*

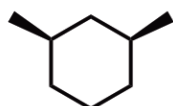


- I-1 a) Draw one version of 1,3-Dimethylcyclohexane that is chiral, and one that is achiral  
b) One of these isomers has two conformers of very different energy. Draw those two chair conformations.

<https://bit.ly/3ATf6gZ>



*chiral*



*achiral*



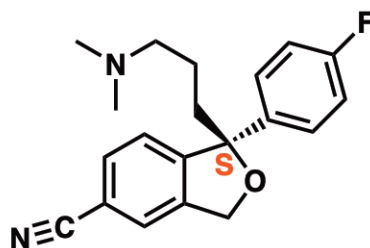
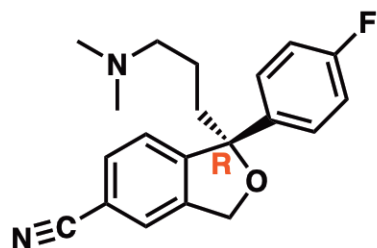
*achiral has conformers of very  
different energy*



### J-1 Draw The Enantiomer (+ more)

<https://bit.ly/3kQpoZW>

Escitalopram (Celexa):



**enantiomer**



• Pure *S* enantiomer shows a specific rotation of  $+120^\circ$ . Sven, a worker in the quality control unit, observed a specific rotation of  $-30^\circ$  for a test sample. What is the percentage of (*R*) and the percentage of (*S*) in that sample?

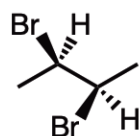
**62.5 % (*R*) 37.5 % (*S*)**

### K-1 Optical Activity

<https://bit.ly/2WI9EV7>

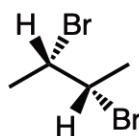
An 80:20 mixture of the (*R,R*) and (*S,S*) enantiomers of 2,3-dibromobutane has an optical rotation of  $-30^\circ$ .

Using these templates, show the stereochemical representation of these compounds, their stereoisomers, and their optical rotations:



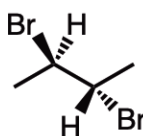
**2*R*, 3*R***

**$[\alpha]: -50^\circ$**



**2*S*, 3*S***

**$[\alpha]: +50^\circ$**



**2*R*, 3*S***

**$[\alpha]: 0^\circ \text{ C}$**



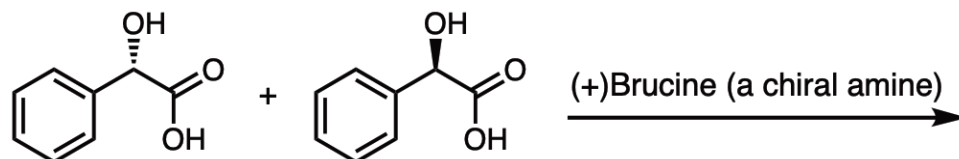


## L-1 Resolution

<https://bit.ly/39IRla2>

Draw the two products of the following reaction, clearly showing stereochemistry (it's OK to use "R<sub>3</sub>N" for (+)-brucine). Note that (+/–) implies a 1:1 mixture of enantiomers.

**Racemic mixture of mandelic acid: reaction with (+)-Brucine**



(+)-mandelic acid    (–)-mandelic acid

- How are these products related to each other? **diastereomers**
- How might you exploit this to resolve mandelic acid into its enantiomers? Describe this process briefly

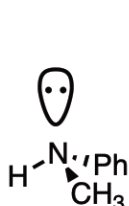
**1) add (+)-brucine to racemic mixture. 2) recrystallize, and separate crystals (one diastereomer) from mother liquor, which contains other diastereomer. 3) add acid to crystals, re-forming mandelic acid, then extract in separatory funnel using water and organic solvent to separate optically active mandelic acid from the salts (can do the same for the mother liquor)**

## M-1 Chiral Nitrogens

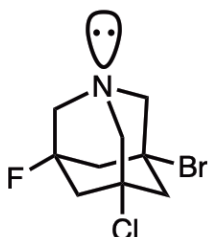
<https://bit.ly/3zNGV99>

Although the nitrogen in the molecule **A** below has four different substituents, the nitrogen does not give rise to a pair of enantiomers. Why not?

Would you expect the nitrogen in molecule **B** to be a chiral center? Why or why not?



**A**



**B**

**constrained geometry; inversion is disfavored.  
nitrogen is a chiral center**

**inverts rapidly, can't be separated**